

9. Conclusions

The kimberlite waters that were sampled show many unusual characteristics compared to most other groundwater studies. The groundwaters in this study have unusually elevated pH and moderately low Eh. A combination of a high pH and Eh values near 0 are consistent with significant water-kimberlite interaction. In conjunction with the pH-K-Mg correlations noted above and the presence of significant dissolved gases (observed during sampling) we suggest that the high pH and low Eh are consistent with low-temperature serpentinization and hydration of ultramafic minerals in the kimberlites.

The $\delta^{2}\text{H}$ and $\delta^{18}\text{O}$ data suggest that most of the kimberlite groundwaters fall within the range of modern meteoric waters. However, there is some question as to why groundwaters from kimberlite A4 do not fall along the OMWL. The A4 waters may reflect fractionation by H_2 and CH_4 production (likely only a very minor component), fractionation as a function of increasing OH^- , and/or mixing with older waters recharged under a different climate. The $\delta^{13}\text{C}$ data are interesting and our initial interpretation is that the process of carbonate dissolution is occurring with an additional source of organic acid likely to explain the most DIC-rich waters.

As this study is still in its infancy, not all data has been acquired and compiled at present. However the available data is compelling and suggests that aqueous geochemistry may be able to be used as a tool for kimberlite exploration. The aqueous geochemistry when compared to the surrounding Archean rock suggests to us that we may be able to use these geochemical signatures to locate kimberlites downflow of kimberlite. The implications of this study is the possibility of a relatively inexpensive and simple method of kimberlite exploration. Aqueous geochemistry, combined with other tried and tested exploration methods, will allow for more efficient exploration and vectoring to targets.

Additional work remaining includes S isotope analyses, completion of trace metal analyses by ICP-MS and radiogenic isotope analyses. It is hoped that the large difference in age and composition between kimberlites and host lithologies will be evident in the isotopic composition of dissolved Pb and Sr.

10. Acknowledgments

This study is being funded as an MSc. thesis project (J. Sader) at the University of Texas at Dallas by the Geological Survey of Canada's Targeted Geoscience Initiative (TGI Project 000022). The Ontario Geological Survey are thanked for providing geological data, field support and helpful advice. P. Sobie of MPH Consulting Ltd. and Sudbury Contact Mines Ltd. are thanked for allowing access to the 95-2 kimberlite and providing geological information. We would also like to thank M. Hinton (GSC) for his helpfulness and time with titration analysis of the samples.

References:

- Barnes, I. and O'Neil, J.R.
1969: The relationship between fluids in some fresh alpine-type ultramafics and possible modern serpentinization, western United States; Geological Society of America Bulletin, v. 80, p. 1947-1960.
- Barnes, I., Rapp, J.B., O'Neil, J.R., Sheppard, R.A., and Gude, A. J. III.
1972: Metamorphic assemblages and the direction of flow of metamorphic fluids in four instances of serpentinization; Contr. Mineral. and Petrol. v. 35, p. 263-276
- Brummer, J.J., MacFadyen, D.A., and Pegg, C.C.
1992a: Discovery of kimberlites in the Kirkland Lake area northern Ontario, Canada, Part I: early surveys and the surficial geology; Exploration Mining Geology, v. 1, No. 4, p. 339-350.
- Brummer, J.J., MacFadyen, D.A., and Pegg, C.C.
1992b: discovery of kimberlites in the Kirkland Lake area northern Ontario, Canada, Part II: kimberlite discoveries, sampling, diamond content, ages and emplacement; Exploration Mining Geology, v. 1, No. 4, p. 351-370.
- Clark, I. and Fritz, P.
1997: Environmental Isotopes in Hydrogeology, Chapter 5; Lewis Publishers, Boca Raton, New York, 328p.
- Deines, P., and Langmuir, D.
1974: Stable carbon isotope ratios and the extent of a gas phase in the evolution of carbonate ground waters; Geochimica et Cosmochimica Acta, v. 38, p. 1147-1164.
- Leybourne, M.I., Goodfellow, W.D., Boyle, D.R., Hall, G.M.
2000: Rapid development of negative Ce anomalies in surface waters and contrasting REE patterns in groundwaters associated with Zn-Pb massive sulfide deposits; Applied Geochemistry, v. 15, p. 695-723.
- McClenaghan, M.B., Kjarsgaard, I.M., Stirling, J.A.R., Pringle, G., Kjarsgaard, B.A. and Berger, B.
1999a: Mineralogy and geochemistry of the C14 kimberlite and associated glacial sediments, Kirkland Lake, Ontario; Geological Survey of Canada, Open File 3719, 147p.
- McClenaghan, M.B., Kjarsgaard, I.M., Kjarsgaard, B.A., Stirling, J.A.R., Pringle, G. and Berger, B.
1999b: Mineralogy and geochemistry of the A4 kimberlite and associated glacial sediments, Kirkland Lake, Ontario; Geological Survey of Canada, Open File 3769, 162p.
- McClenaghan, M.B., Kjarsgaard, I.M., Schulze, D.J., Stirling, J.A.R., Berger, B., and Pringle, G.
1998: Mineralogy and geochemistry of the Diamond Lake kimberlite and associated esker sediments, Kirkland Lake, Ontario; Geological Survey of Canada, open File 3576, 200p.
- McClenaghan, M.B., Kjarsgaard, I.M., Schulze, D.J., Stirling, J.A.R., Pringle, G., and Berger, B.
1996: Mineralogy and geochemistry of the B30 kimberlite and overlying glacial sediments, Kirkland Lake, Ontario; Geological Survey of Canada, Open File 3295, 245p.
- Sage, R.P.
1996: Kimberlites of the Lake Timiskaming structural zone; Ontario Geological Survey Open file Report 5937, 435 p.
- Sherwood Lollar, B., Frapre, S.K., Weise, S.M., Fritz, P., Macko, A., and Welhan, J.A.
1993: Abiogenic methanogenesis in crystalline rocks; Geochimica et Cosmochimica Acta, v. 57, p. 5087-5097.