

Transport Canada Safety and Security

Transports Canada Sécurité et sûreté

Road Safety

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**Standards and Regulations Division** 

## **TEST METHOD SECTION 116 Hydraulic Brake Fluids**

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Motor Vehicle Standards and Research Branch Road Safety and Motor Vehicle Regulation Directorate TRANSPORT CANADA Ottawa, Ontario K1A 0N5

## TEST METHOD SECTION 116 — HYDRAULIC BRAKE FLUIDS

#### **1.** INTRODUCTION

Subsections 2, to 4, Figures 1 to 6, and Tables I to IV of this section make up test methods referred to in section 116 of Schedule IV to the Motor Vehicle Safety Regulations, to demonstrate compliance with the requirements of section 116 of Schedule IV of the said Regulations.

(Original signed by)

Harvey J. Layden for the Minister of Transport Ottawa, Ontario

## **2. DEFINITIONS**

In this Section,

"duplicate samples" means two samples of brake fluid taken from a single packaged lot and tested simultaneously;

"silicone base brake fluid" (SBBF) means a brake fluid which consists of not less than 70 percent by weight of a diorgano polysiloxane;

"TEGME" means SAE triethylene glycol monomethyl ether, brake fluid grade, referee material (TEGME) as described in Appendix E of SAE Standard J1703 NOV83, "Motor Vehicle Brake Fluid," November 1983.

## 2.1 CROSS REFERENCES

The following table cross-references the performance requirements of section 116 of the Motor Vehicle Safety Regulations and the appropriate test procedures of this test method.

	TEST METHOD	REQUIREMENTS
	PROCEDURES	CMVSS 116
Equilibrium Reflux Boiling Point	3.1	(2)(a), (3)(a), (4)(a)
Wet ERBP	3.2	(2)(b), (3)(b), (4)(b)
Kinematic Viscosity	3.3	(2)(c), (3)(c), (4)(c)
pH Value	3.4	(5)(a)
High Temperature Stability	3.5.3	(5)(b)
Chemical Stability	3.5.4	(5)(c)
Corrosion	3.6	(5)(d)
Fluidity and Appearance	3.7	(5)(e)
Evaporation	3.8	(5)(f)
Water Tolerance	3.9	(5)(g)
Compatibility	3.10	(5)(h)
Resistance to Oxidation	3.11	(5)(i)
Effect on Rubber Caps	3.12	(5)(j)
Stroking Properties	3.13	(5)(k)

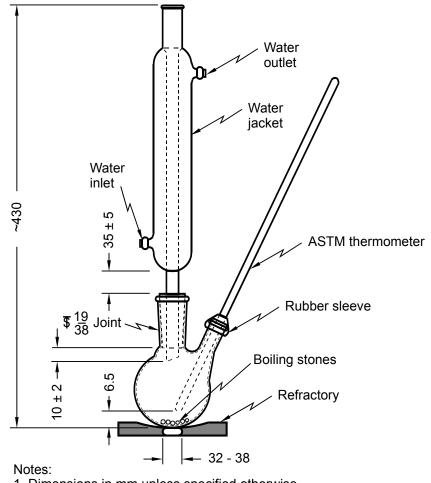
#### **3. TEST PROCEDURES**

## 3.1 EQUILIBRIUM REFLUX BOILING POINT

Determine the ERBP of a brake fluid by running duplicate samples according to the following procedure and averaging the results.

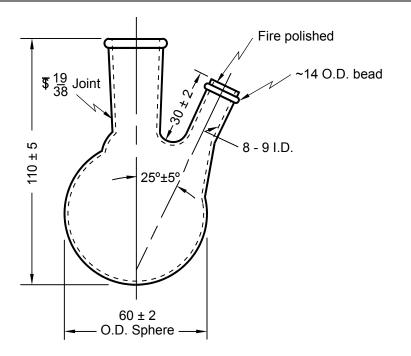
- 3.1.1 Summary of Procedure Sixty millilitres (mL) of brake fluid are boiled under specified equilibrium conditions (reflux) at atmospheric pressure in a 100 mL flask. The average temperature of the boiling fluid at the end of the reflux period, corrected for variations in barometric pressure if necessary, is the ERBP.
- 3.1.2 Apparatus (see Figure 1) The test apparatus shall consist of:
  - a) flask (see Figure 2) a 100 mL round-bottom, short-neck heatresistant glass flask having a neck with a 19/38 standard taper, female ground-glass joint and a side-entering tube, with an outside diameter of 10 millimetres (mm) which centers the thermometer bulb in the flask 6.5 mm from the bottom;
    - b) condenser a water-cooled, reflux, glass-tube type, condenser having a jacket 200 mm in length, the bottom end of which has a 19/38 standard-taper, drip-tip, male ground-glass joint;
    - c) boiling stones three clean, unused silicon carbide grains (approximately 2 mm in diameter, grit No. 8);
  - d) thermometer standardized calibrated partial immersion (76 mm), solid stem, thermometers conforming to the requirements for an ASTM 2C or 2F, and an ASTM 3C or 3F thermometer; and
  - e) heat source variable autotransformer-controlled heating mantle designed to fit the flask, or an electric heater with rheostat heat control.
- 3.1.3 Preparation of Apparatus
  - a) Thoroughly clean and dry all glassware.
  - b) Insert thermometer through the side tube until the tip of the bulb is
    6.5 mm from the bottom centre of the flask. Seal with a short piece of natural rubber, EPDM, SBR, or butyl tubing.
  - c) Place  $60 \pm 1$  ml of brake fluid and the silicon carbide grains into the flask.

- d) Attach the flask to the condenser. When using a heating mantle, place the mantle under the flask and support it with a ring-clamp and laboratory-type stand, holding the entire assembly in place by a clamp. When using a rheostat-controlled heater, center a standard porcelain or hard asbestos refractory, having an opening of 32 to 38 mm diameter, over the heating element and mount the flask so that direct heat is applied only through the opening in the refractory. Place the assembly in an area free from drafts or other types of sudden temperature changes. Connect the cooling water inlet and outlet tubes to the condenser. Turn on the cooling water. The water supply temperature shall not exceed 28°C and the temperature rise through the condenser shall not exceed 2°C.
- 3.1.4 Procedure Apply heat to the flask so that within  $10 \pm 2$  minutes the fluid is refluxing in excess of 1 drop per second. The reflux rate shall not exceed 5 drops per second at any time. Immediately adjust the heating rate to obtain an equilibrium reflux rate of 1 to 2 drops per second over the next  $5 \pm 2$  minutes. Maintain this rate for an additional 2 minutes, taking four temperature readings at 30second intervals. Record the average of these as the observed ERBP. If no reflux is evident when the fluid temperature reaches  $260^{\circ}$ C, discontinue heating and report the ERBP as being in excess of  $260^{\circ}$ C.



- 1. Dimensions in mm unless specified otherwise.
- 2. Not to scale

Figure 1 – Boiling Point Test Apparatus



Notes:

- 1. Dimensions in mm unless specified otherwise.
- 2. Not to scale

#### Figure 2 – Detail of 100 mL Short-Neck Flask

- 3.1.5 Calculation
  - a) Thermometer inaccuracy. Correct the observed ERBP by applying any correction factor obtained in standardizing the thermometer.
  - b) Variation from standard barometric pressure. Apply the factor shown in Table I to calculate the barometric pressure correction to the ERBP.

#### TABLE I — Correction For Barometric Pressure

Observed ERBP Corrected for thermometer inaccuracy	Correction per 1 mm Difference in Pressure * °C
100 <sup>°</sup> C to 190 <sup>°</sup> C	0.039
Over 190 °C	0.04

To be added in case barometric pressure is below 760 mm; to be subtracted in case barometric pressure is above 760 mm.

c) If the two corrected observed ERBPs agree within 2°C (4°C for brake fluids having an ERBP over 230°C) average the duplicate runs as the ERBP; otherwise, repeat the entire test, averaging the four corrected observed values to determine the original ERBP.

## 3.2 WET ERBP

Determine the wet ERBP of a brake fluid by running duplicate samples according to the following procedure.

- 3.2.1 Summary of the Procedure A 350 mL sample of the brake fluid is humidified under controlled conditions; 350 mL of SAE triethylene glycol monomethyl ether, brake fluid grade, referee material (TEGME) as described in Appendix E of SAE Standard J1703 NOV83, "Motor Vehicle Brake Fluid," November 1983, is used to establish the end point for humidification. After humidification the water content and ERBP of the brake fluid are determined.
- 3.2.2 Apparatus for Humidification (see Figure 3) The test apparatus shall consist of:
  - a) glass jars four SAE RM-49 corrosion test jars or equivalent screw-top, straight-sided, round glass jars each having a capacity of about 475 mL and approximate inner dimensions of 100 mm in height by 75 mm in diameter, with matching lids having new, clean inserts providing water-vapour-proof seals;
  - b) desiccator and cover two bowl-form glass desiccators, 250 mm inside diameter, having matching tubulated covers fitted with No. 8 rubber stoppers; and
  - c) desiccator plate two 230 mm diameter, perforated porcelain desiccator plates, without feet, glazed on one side.
- 3.2.3 Reagents and Materials
  - a) Distilled water, see 4.1.
  - b) SAE TEGME referee material, see 3.2.1.
- 3.2.4 Preparation of Apparatus Lubricate the ground-glass joint of the desiccator. Pour  $450 \pm 10$  mL of distilled water into each desiccator and insert perforated porcelain desiccator plates. Place the desiccators in an oven with temperature controlled at  $50 \pm 1^{\circ}$ C throughout the humidification procedure.

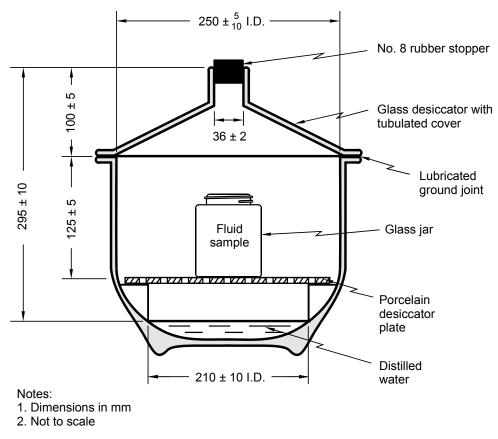


Figure 3 – Humidification Apparatus

3.2.5

Procedure - Pour  $350 \pm 5$  mL of the brake fluid into an open corrosion test jar. Prepare in the same manner a duplicate test fluid sample, and two duplicate specimens of the SAE TEGME referee material ( $350 \pm 5 \text{ mL of TEGME in each jar}$ ). Adjust the water content of the SAE TEGME fluid to  $0.50 \pm 0.05$  percent by weight at the start of the test in accordance with 4.2. Place one sample each of the test brake fluid and the prepared TEGME sample into the same desiccator. Repeat for the second sample of test brake fluid and TEGME sample in a second desiccator. Place the desiccators in the 50°C controlled oven and replace the desiccator covers. At intervals during oven humidification, remove the rubber stoppers in the tops of desiccators. Using a long needled hypodermic syringe, take a sample of not more than 2 mL from each TEGME sample and determine its water content. Remove no more than 10 mL of fluid from each SAE TEGME fluid sample during the humidification procedure. When the water content of the SAE TEGME fluid sample reaches  $3.70 \pm$ 0.05 percent by weight (averages of the duplicates), remove the two test fluid specimens from their desiccators and promptly cap each jar tightly. Allow the sealed jars to cool for 60 to 90 minutes

at  $23 \pm 5^{\circ}$ C. Measure the water content of the test fluid specimens in accordance with 4.2 and determine their ERBPs in accordance with 3.1. If the two ERBPs agree within 4°C, average them to determine the wet ERBP; otherwise repeat and average the four individual ERBPs as the wet ERBP of the brake fluid.

## 3.3 KINEMATIC VISCOSITY

Determine the kinematic viscosity of a brake fluid in square millimetres per second  $(mm^2/s)$  by the following procedure. Run duplicate samples at each of the specified temperatures, making two timed runs on each sample.

- 3.3.1 Summary of the Procedure The time is measured for a fixed volume of the brake fluid to flow through a calibrated glass capillary viscometer under an accurately reproducible head and at a closely controlled temperature. The kinematic viscosity is then calculated from the measured flow time and the calibration constant of the viscometer.
- 3.3.2 Apparatus
  - a) Viscometers. Calibrated glass capillary-type viscometers, conforming to ASTM standard D2515-66, "Standard Specification for Kinematic Glass Viscometers", measuring viscosity within the precision limits of 3.4.7. Use suspended level viscometers for viscosity measurements at low temperatures. Use Cannon-Fenske Routine or other modified Ostwald viscometers at ambient temperatures and above.
  - b) Viscometer holders and frames. Mount a viscometer in the constant-temperature bath so that the mounting tube is held within  $1^{\circ}$  of the vertical.
  - c) Viscometer bath. A transparent liquid bath of sufficient depth such that at no time during the measurement will any portion of the sample in the viscometer be less than 2 cm below the surface or less than 2 cm above the bottom. The bath shall be cylindrical in shape, with turbulent agitation sufficient to meet the temperature control requirements. For measurements within 15°C to 100°C the temperature of the bath medium shall not vary by more than 0.01°C over the length of the viscometers, or between the positions of the viscometers, or at the locations of the thermometers. Outside this range, the variation shall not exceed 0.03°C.
  - d) Thermometers. Liquid-in-Glass Kinematic Viscosity Test Thermometers, covering the range of test temperatures indicated in

Table II and conforming to ASTM standard E1-68, "Specifications for ASTM Thermometers," and in the IP requirements for IP Standard Thermometers. Standardize before use (see 3.3.3(b)). Use two standardized thermometers in the bath.

Temperature range	For tests at	Subdivisions		nometer Jo.
°C	°C	°C	ASTM	IP
-55.3 to - 52.5	-55	0.05	74 F	69 F or C
-41.4 to - 38.6	-40	0.05	73 F	68 F or C
98.6 to 101.4	100	0.05	30 F	32 F or C

#### TABLE II — Kinematic Viscosity Thermometers

e) Timing device. Stop watch or other timing device graduated in divisions representing not more than 0.2 second, with an accuracy of at least  $\pm$  0.05 percent when tested over intervals of 15 minutes. Electrical timing devices may be used when the current frequency is controlled to an accuracy of 0.01 percent or better.

#### 3.3.3 Standardization

a) Viscometers. Use viscometers calibrated in accordance with Appendix 1 of ASTM standard D445-65, "Viscosity of Transparent and Opaque Liquids (Kinematic and Dynamic Viscosities)". The calibration constant, C, is dependent upon the gravitational acceleration at the place of calibration. This must, therefore, be supplied by the standardization laboratory together with the instrument constant.

Where the acceleration of gravity, g, in the two locations differs by more than 0.1 percent, correct the calibration constant as follows:

$$C_2 = (g_2/g_1) \times C_1$$

where the subscripts 1 and 2 indicate respectively the standardization laboratory and the testing laboratory.

b) Thermometers. Check liquid-in-glass thermometers to the nearest 0.01°C by direct comparison with a standardized thermometer. Kinematic Viscosity Test Thermometers shall be standardized at "total immersion". The ice point of standardized thermometers shall be determined before use and the official corrections shall be adjusted to conform to the changes in ice points. (See ASTM

standard E77-66, "Verification and Calibration of Liquid-in-Glass Thermometers.")

- c) Timers. Time signals are broadcast by Station CHU from Ottawa, Canada, at 3.330, 7.335, and 14.670 MHz. Time signals are also broadcast by the National Bureau of Standards, Station WWV, Washington, D.C. at 2.5, 5, 10, 15, 20, 25, 30 and 35 MHz, and Station MSF at Rugby, United Kingdom, at 2.5, 5, and 10 MHz.
- 3.3.4 Procedure
  - a) Set and maintain the bath at the appropriate test temperature within the limits specified in 3.3.2(c). Apply the necessary corrections, if any, to all thermometer readings.
  - b) Select a clean, dry, calibrated viscometer giving a flow time not less than its specified minimum, or 200 seconds, whichever is the greater.
  - c) Charge the viscometer in the manner used when the instrument was calibrated. Do not filter or dry the brake fluid, but protect it from contamination by dirt and moisture during filling and measurements.
    - (i) Charge the suspended level viscometers by tilting about 30° from the vertical and pouring sufficient brake fluid through the fill tube into the lower reservoir so that when the viscometer is returned to vertical position the meniscus is between the fill marks. For measurements below 0°C, before placing the filled viscometer into the constant temperature bath, draw the sample into the working capillary and timing bulb and insert small rubber stoppers to suspend the fluid in this position, to prevent accumulation of water condensate on the walls of the critical portions of the viscometer. Alternatively, fit loosely packed drying tubes into the open ends of the viscometer to prevent water condensation, but do not restrict the flow of the sample under test by the pressure created in the instrument.
    - (ii) If a Cannon-Fenske Routine viscometer is used, charge by inverting and immersing the smaller arm into the brake fluid and applying vacuum to the larger arm. Fill the tube to the upper timing mark, and return the viscometer to an upright position.
  - d) Mount the viscometer in the bath in a true vertical position (see 3.3.2(b)).

- e) The viscometer shall remain in the bath until it reaches the test temperature.
- f) At temperatures below  $0^{\circ}$ C conduct an untimed preliminary run by allowing the brake fluid to drain through the capillary into the lower reservoir after the test temperature has been established.
- g) Adjust the head level of the brake fluid to a position in the capillary arm about 5 mm above the first timing mark.
- h) With brake fluid flowing freely measure to within 0.2 second the time required for the meniscus to pass from the first timing mark to the second. If this flow time is less than the minimum specified for the viscometer or 200 seconds, whichever is greater, repeat using a viscometer with a capillary of smaller diameter.
- i) Repeat 3.3.4(g) and (h). If the two timed runs do not agree within 0.2 percent, reject and repeat using a fresh sample of brake fluid.
- 3.3.5 Cleaning the Viscometers
  - a) Periodically clean the instrument with chromic acid to remove organic deposits. Rinse thoroughly with distilled water and acetone, and dry with clean dry air.
  - b) Between successive samples, rinse the viscometer with ethanol (isopropanol when testing DOT 5 SBBFs) followed by an acetone or ether rinse. Pass a slow stream of filtered dry air through the viscometer until the last trace of solvent is removed.
- 3.3.6 Calculation
  - a) The following viscometers have a fixed volume charged at ambient temperature, and as a consequence C varies with test temperature: Cannon-Fenske Routine, Pinkevitch, Cannon-Manning Semi-Micro, and Cannon-Fenske Opaque. To calculate C at test temperatures other than the calibration temperatures for these viscometers, see ASTM standard D2515-66, "Kinematic Glass Viscometers" or follow instructions given on the manufacturer's certificate of calibration.
  - b) Average the four timed runs on the duplicate samples to determine the kinematic viscosities.

3.3.7 Precision (at 95 percent confidence level)

Repeatability. If the results on duplicate samples by the same operator differ by more than 1 percent of their mean, repeat the tests.

## 3.4 PH VALUE

Determine the pH value of a brake fluid by running one sample according to the following procedure.

- 3.4.1 Summary of the Procedure Brake fluid is diluted with an equal volume of an ethanol-water solution. The pH of the resultant mixture is measured with a prescribed pH meter assembly at 23°C.
- 3.4.2 Apparatus The pH assembly consists of the pH meter, glass electrode, and calomel electrode, as specified in Appendices A1.1, A1.2 and A1.3 of ASTM standard D1121-67, "Standard Method of Test for Reserve Alkalinity of Engine Antifreezes and Antirusts". The glass electrode is a full range type (pH 0-14), with low sodium error.
- 3.4.3 Reagents Reagent grade chemicals conforming to the specifications of the Committee on Analytical Reagents of the American Chemical Society.
  - a) Distilled Water. Distilled water (see section 4.1) shall be boiled for about 15 minutes to remove carbon dioxide, and protected with a soda-lime tube or its equivalent while cooling and in storage. (Take precautions to prevent contamination by the materials used for protection against carbon dioxide). The pH of the boiled distilled water shall be between 6.2 and 7.2 at 25°C.
  - b) Standard buffer solutions. Prepare buffer solutions for calibrating the pH meter and electrode pair from salts sold specifically for use, either singly or in combination, as pH standards. Dry salts for 1 hour at 110°C before use except for borax which shall be used as the decahydrate. Store solutions with pH less than 9.5 in bottles of chemically resistant glass or polyethylene. Store the alkaline phosphate solution in a glass bottle coated inside with paraffin. Do not use a standard with an age exceeding three months.
    - (i) Potassium hydrogen phthalate buffer solution (0.05 M, pH = 4.01 at 25°C). Dissolve 10.21 g of potassium hydrogen phthalate (KHC<sub>8</sub>H<sub>4</sub>0<sub>4</sub>) in distilled water. Dilute to 1 litre.

- (ii) Neutral phosphate buffer solution (0.025 M with respect to each phosphate salt, pH = 6.86 at 25°C). Dissolve 3.40 g of potassium dihydrogen phosphate (KH<sub>2</sub>PO<sub>4</sub>) and 3.55 g of anhydrous disodium hydrogen phosphate (Na<sub>2</sub>HPO<sub>4</sub>) in distilled water.
- (iii) Borax buffer solution (0.01 M, pH =  $9.18 \text{ at } 25^{\circ}\text{C}$ ). Dissolve 3.81 g of disodium tetraborate decahydrate (Na<sub>2</sub>B<sub>4</sub>O<sub>7</sub>·10 H<sub>2</sub>O) in distilled water, and dilute to 1 litre. Stopper the bottle except when actually in use.
- (iv) Alkaline phosphate buffer solution (0.01 M, trisodium phosphate,  $pH = 11.72 \text{ at } 25^{\circ}\text{C}$ ). Dissolve 1.42 g of anhydrous disodium hydrogen phosphate (Na<sub>2</sub>HPO<sub>4</sub>) in 100 mL of a 0.1 M carbonate-free solution of sodium hydroxide. Dilute to 1 litre with distilled water.
- (v) Potassium chloride electrolyte. Prepare a saturated solution of potassium chloride (KCl) in distilled water.
- c) Ethanol-water mixture. To 80 parts by volume of ethanol (see section 4.3) add 20 parts by volume of distilled water. Adjust the pH of the mixture to  $7 \pm 0.1$  using 0.1 N sodium hydroxide (NaOH) solution. If more than 4 mL of NaOH solution per litre of mixture is required for neutralization, discard the mixture.
- 3.4.4 Preparation of Electrode System
  - a) Maintenance of electrodes. Clean the glass electrode before using by immersing in cold chromic-acid cleaning solution. Drain the calomel electrode and fill with KCl electrolyte, keeping level above that of the mixture at all times. When not in use, immerse the lower halves of the electrodes in distilled water, and do not immerse in the mixture for any appreciable period of time between determinations.
  - b) Preparation of electrodes. Condition new glass electrodes and those that have been stored dry as recommended by the manufacturer. Before and after using, wipe the glass electrode thoroughly with a clean cloth, or a soft absorbent tissue, and rinse with distilled water. Before each pH determination, soak the prepared electrode in distilled water for at least 2 minutes. Immediately before use, remove any excess water from the tips of the electrode.

- 3.4.5 Standardization of the pH Assembly and Testing of the Electrodes
  - a) Immediately before use, standardize the pH assembly with a standard buffer solution. Then use a second standard buffer solution to check the linearity of the response of the electrodes at different pH values, and to detect a faulty glass electrode or incorrect temperature compensation. The two buffer solutions bracket the anticipated pH value of the test brake fluid.
  - b) Allow instrument to warm up, and adjust according to the manufacturer's instructions. Immerse the tips of the electrodes in a standard buffer solution and allow the temperature of the buffer solution and the electrodes to equalize. Set the temperature knob at the temperature of the buffer solution. Adjust the standardization of asymmetry potential control until the meter registers a scale reading, in pH units, equal to the known pH of the standardizing buffer solution.
  - c) Rinse the electrodes with distilled water and remove excess water from the tips. Immerse the electrodes in a second standard buffer solution. The reading of the meter shall agree with the known pH of the second standard buffer solution within  $\pm$  0.05 unit without changing the setting of the standardization of asymmetry potential control.
  - d) A faulty electrode is indicated by failure to obtain a correct value for the pH of the second standard buffer solution after the meter has been standardized with the first.
- 3.4.6 Procedure To  $50 \pm 1$  mL of the test brake fluid add  $50 \pm 1$  mL of the ethanol-water mixture (see 3.4.3(c)) and mix thoroughly. Immerse the electrodes in the mixture. Allow the system to come to equilibrium, re-adjust the temperature compensation if necessary, and take the pH reading.

#### **3.5 FLUID STABILITY**

Evaluate the heat and chemical stability of a brake fluid by the following procedure, running duplicate samples for each test and averaging the results.

3.5.1 Summary of the Procedure - The degradation of the brake fluid at elevated temperature, alone or in a mixture with a reference fluid, is evaluated by determining the change in boiling point after a period of heating under reflux conditions.

- 3.5.2 Apparatus Use the apparatus and preparation specified in 3.1.2 and 3.1.3.
- 3.5.3 High Temperature Stability
  - 3.5.3.1 Procedure Heat a new  $60 \pm 1$  mL sample of the brake fluid to  $185 \pm 2^{\circ}$ C. Hold at this temperature for  $120 \pm 5$  minutes. Bring to a reflux rate in excess of 1 drop per second within 5 minutes. The reflux rate should not exceed 5 drops per second at any time. Over the next  $5 \pm 2$  minutes adjust the heating rate to obtain an equilibrium reflux rate of 1 to 2 drops per second. Maintain this rate for an additional 2 minutes, taking four temperature readings at 30-second intervals. Average these as the observed ERBP. If no reflux is evident when the fluid temperature reaches  $260^{\circ}$ C discontinue heating and report ERBP as in excess of  $260^{\circ}$ C.
  - 3.5.3.2 Calculation Correct the observed ERBP for thermometer and barometric pressure factors according to 3.1.5(a) and (b). Average the corrected ERBPs of the duplicate samples. The difference between this average and the original ERBP obtained in 3.1 is the change in ERBP of the fluid.
- 3.5.4 Chemical Stability
  - 3.5.4.1 Materials SAE Compatibility Fluid.
  - 3.5.4.2 Procedure
    - a) Mix  $30 \pm 1$  mL of the brake fluid with  $30 \pm 1$  mL of SAE Compatibility Fluid in a boiling point flask (3.1.2(a)). Determine the initial ERBP of the mixture by applying heat to the flask so that the fluid is refluxing in  $10 \pm 2$  minutes at a rate in excess of 1 drop per second, but not more than 5 drops per second. Note the maximum fluid temperature observed during the first minute after the fluid begins refluxing at a rate in excess of 1 drop per second. Over the next  $15 \pm 1$  minutes, adjust and maintain the reflux rate at 1 to 2 drops per second. Maintain this rate for an additional 2 minutes, recording the average value of four temperature readings taken at 30-second intervals as the final ERBP.
    - b) Thermometer and barometric corrections are not required.
  - 3.5.4.3 Calculation The difference between the initial ERBP and the final average temperature is the change in temperature of the refluxing mixture. Average the results of the duplicates to the nearest 0.5°C.

### **3.6** CORROSION

Evaluate the corrosiveness of a brake fluid by running duplicate samples according to the following procedure.

- 3.6.1 Summary of the Procedure Six specified metal corrosion test strips are polished, cleaned and weighed, then assembled as described. Assembly is placed on a standard wheel cylinder cup in a corrosion test jar, immersed in the water-wet brake fluid, capped and placed in an oven at 100°C for 120 hours. Upon removal and cooling, the strips, fluid, and cups are examined and tested.
- 3.6.2 Equipment
  - a) Balance. An analytical balance having a minimum capacity of 50 g and capable of weighing to the nearest 0.1 mg.
  - b) Desiccators. Desiccators containing silica gel or other suitable desiccant.
  - c) Oven. Gravity convection oven capable of maintaining the desired set point within 2°C.
  - d) Micrometer. A machinist's micrometer, 25 to 50 mm capacity, or an optical comparator, capable of measuring the diameter of the SBR wheel cylinder (WC) cups to the nearest 0.02 mm.
- 3.6.3 Materials
  - a) Corrosion test strips. Two sets of strips from each of the metals listed in Appendix C of SAE Standard J1703 NOV83, "Motor Vehicle Brake Fluid," November 1983. Each strip shall be approximately 8 cm long, 1.3 cm wide, not more than 0.6 cm thick, and have a surface area of  $25 \pm 5$  sq. cm and a hole 4 to 5 mm in diameter on the centreline about 6 mm from one end. The hole shall be clean and free from burrs. Tinned iron strips shall be unused. Other strips, if used, shall not be employed if they cannot be polished to a high finish.
  - b) SBR cups. Two unused standard SAE SBR wheel cylinder (WC) cups, as specified in 4.6.
  - c) Corrosion test jars and lids. Two screw-top straight-sided round glass jars, each having a capacity of approximately 475 mL and inner dimensions of approximately 100 mm in height and 75 mm

in diameter, and a tinned steel lid (no insert or organic coating) vented with a hole  $0.8 \pm 0.1$  mm in diameter (No. 68 drill).

- d) Machine screws and nuts. Clean, rust and oil-free, uncoated mild steel round or fillister head machine screws, size 6 or 8 32 UNC-Class 2A, 5/8 or 3/4 inch long (or equivalent metric sizes), and matching uncoated nuts.
- e) Supplies for polishing strips. Waterproof silicon carbide paper, grit No. 320 A; grade 00 steel wool, lint-free polishing cloth.
- f) Distilled water as specified in 4.1.
- g) Ethanol as specified in 4.3.
- h) Isopropanol as specified in 4.7.
- 3.6.4 Preparation
  - Corrosion test strips. Except for the tinned iron strips, abrade a) corrosion test strips on all surface areas with silicon carbide paper wet with ethanol (isopropanol when testing DOT 5 SBBFs) until all surface scratches, cuts and pits are removed. Use a new piece of paper for each different type of metal. Polish the strips with the 00 grade steel wool. Wash all strips, including the tinned iron and the assembly hardware, with ethanol (isopropanol when testing DOT 5 SBBFs); dry the strips and assembly hardware with a clean lint free cloth or use filtered compressed air and place the strips and hardware in a desiccator containing silica gel or other suitable desiccant and maintained at  $23 \pm 5$ °C, for at least 1 hour. Handle the strips with forceps after polishing. Weigh and record the weight of each strip to the nearest 0.1 mg. Assemble the strips on a clean dry machine screw, with matching plain nut, in the order of tinned iron, steel, aluminum, cast iron, brass, and copper. Bend the strips other than the cast iron, so that there is a separation of  $3 \pm$ <sup>1</sup>/<sub>2</sub> mm between adjacent strips for a distance of about 5 cm from the free end of the strips. (See Figure 4.) Tighten the screw on each test strip assembly so that the strips are in electrolytic contact, and can be lifted by either of the outer strips (tinned iron or copper) without any of the strips moving relative to the others when held horizontally. Immerse the strip assemblies in 90 percent ethyl alcohol. Dry with dried filtered compressed air, then desiccate at least 1 hour before use.
  - b) SBR WC cups. Measure the base diameters of the two standard SBR cups, using an optical comparator or micrometer, to the nearest 0.02 mm along the centreline of the SAE and rubber-type

identifications and at right angles to this centreline. Take the measurements at least 0.4 mm above the bottom edge and parallel to the base of the cup. Discard any cup if the two measured diameters differ by more than 0.08 mm. Average the two readings on each cup. Determine the hardness of the cups according to 4.4.

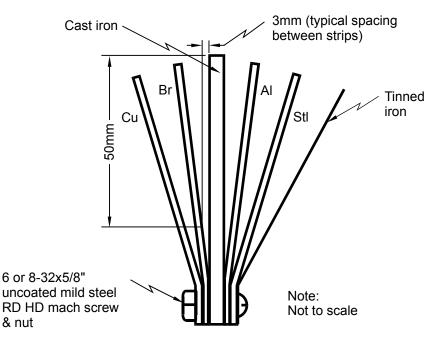
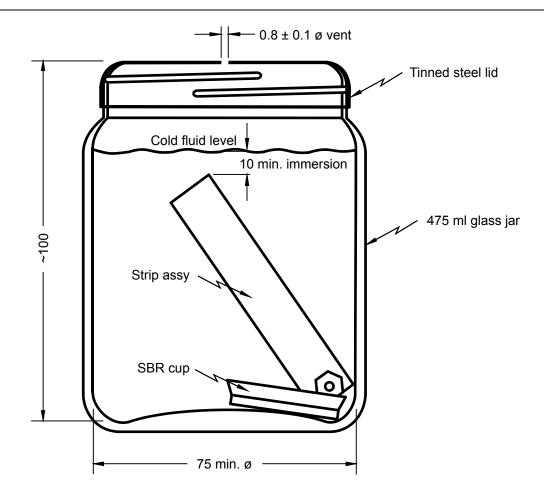


Figure 4 – Corrosion Strip Assembly

3.6.5

Procedure - Rinse the cups in ethanol (isopropanol when testing DOT 5 SBBFs) for not more than 30 seconds and wipe dry with a clean lint-free cloth. Place one cup with lip edge facing up, in each jar. Insert a metal strip assembly inside each cup, with the fastened end down and the free end extending upward (see Figure 5). When testing brake fluids, except DOT 5 SBBFs, mix 760 mL of brake fluid with 40 mL of distilled water. When testing DOT 5 SBBFs, humidify 800 mL of brake fluid in accordance with 3.2, eliminating determination of the ERBP. Using this water wet mixture, cover each strip assembly to a minimum depth of 10 mm above the tops of the strips. Tighten the lids and place the jars for  $120 \pm 2$  hours in an oven maintained at  $100 \pm 2$ °C. Allow the jars to cool at  $23 \pm 5^{\circ}$ C for 60 to 90 minutes. Immediately remove the strips from the jars using forceps, agitating the strip assembly in the fluid to remove loose adhering sediment. Examine the test strips and jars for adhering crystalline deposits. Disassemble the metal strips, and remove adhering fluid by flushing with water; clean each strip by wiping with a clean cloth wetted with ethanol (isopropanol when testing DOT 5 SBBFs). Examine the strips for evidence of corrosion and pitting. Disregard staining or

discoloration. Place the strips in a desiccator containing silica gel or other suitable desiccant, maintained at  $23 \pm 5$  °C, for at least 1 hour. Weigh each strip to the nearest 0.1 mg. Determine the change in weight of each metal strip. Average the results for the two strips of each type of metal. Immediately following the cooling period, remove the cups from the jars with forceps. Remove loose adhering sediment by agitation of the cups in the mixture. Rinse the cups in ethanol (isopropanol when testing DOT 5 SBBFs) and air-dry. Examine the cups for evidence of sloughing, blisters, and other forms of disintegration. Measure the base diameter and hardness of each cup within 15 minutes after removal from the mixture. Examine the mixture for gelling. Agitate the mixture to suspend and uniformly disperse sediment. From each jar, transfer one 100 mL portion of the mixture to an ASTM cone-shaped centrifuge tube. Determine the percent sediment after centrifuging as described in 4.5. Measure the pH value of the corrosion test fluid according to 3.4.6. Measure the pH value of the test mixture according to 3.4.6.



Notes:

1. Dimensions in mm unless specified otherwise.

2. Not to scale

#### FIGURE 5 – Corrosion Test Apparatus

#### 3.6.6 Calculation

- a) Measure the area of each type of test strip to the nearest square centimetre. Divide the average change in weight for each type by the area of that type.
- b) Note other data and evaluations indicating compliance with section 116 of the said Regulations. In the event of a marginal pass on inspection by attributes, or of a failure in one of the duplicates, run another set of duplicate samples. Both repeat samples shall meet all requirements of section 116 of the said Regulations.

#### **3.7** FLUIDITY AND APPEARANCE AT LOW TEMPERATURES

Determine the fluidity and appearance of a sample of brake fluid at each of two selected temperatures by the following procedure.

- 3.7.1 Summary of Procedure Brake fluid is chilled to expected minimum exposure temperatures and observed for clarity, gellation, sediment, separation of components, excessive viscosity or thixotropy.
- 3.7.2 Apparatus
  - a) Oil sample bottle. Two clear flint glass 4-ounce bottles made especially for sampling oil and other liquids, with a capacity of approximately 125 mL, an outside diameter of  $37 \pm 0.05$  mm and an overall height of  $165 \pm 2.5$  mm.
  - b) Cold chamber. An air bath cold chamber capable of maintaining storage temperatures down to  $-55^{\circ}$ C with an accuracy of  $\pm 2^{\circ}$ C.
  - c) Timing device. A timing device in accordance with 3.3.2(e).
- 3.7.3 Procedure
  - a) Place  $100 \pm 1$  mL of brake fluid at room temperature in an oil sample bottle. Stopper the bottle with an unused cork and place in the cold chamber at the higher storage temperature specified in the Table referred to in paragraph (5)(e) of section 116 of the said Regulations. After  $144 \pm 4$  hours remove the bottle from the chamber, quickly wipe with a clean, lint-free cloth saturated with ethanol (isopropanol when testing DOT 5 SBBFs) or acetone. Examine the fluid for evidence of sludging, crystallization, stratification or sedimentation. Invert the bottle and determine the number of seconds required for the air bubble to travel to the top of the fluid. Let the sample warm to room temperature and then examine it.
  - b) Repeat 3.7.3(a), substituting the lower cold chamber temperature specified in the Table referred to in paragraph (5)(e) of section 116 of the said Regulations, and a storage period of 6 hours  $\pm$  12 minutes.
  - **NOTE**: Test specimens from either storage temperature may be used for the other only after warming up to room temperature.

#### **3.8 EVAPORATION**

The evaporation residue, and pour point of the evaporation residue of brake fluid, are determined by the following procedure. Four replicate samples are run.

- 3.8.1 Summary of the Procedure The volatile diluent portion of a brake fluid is evaporated in an oven at  $100^{\circ}$ C. The nonvolatile lubricant portion (evaporation residue) is measured and examined for grittiness; the residues are then combined and checked to assure fluidity at -5°C.
- 3.8.2 Apparatus
  - a) Petri dishes. Four covered glass petri dishes approximately 100 mm in diameter and 15 mm in height.
  - b) Oven. A top-vented gravity-convection oven capable of maintaining a temperature of  $100 \pm 2^{\circ}$ C.
  - c) Balance. A balance having a capacity of at least 100 g, capable of weighing to the nearest 0.01 g, and suitable for weighing the petri dishes.
  - d) Oil sample bottle. A glass sample bottle as described in 3.7.2(a).
  - e) Cold chamber. Air bath cold chamber capable of maintaining an oil sample bottle at  $-5 \pm 1^{\circ}$ C.
  - f) Timing device. A timing device as described in 3.3.2(e).
- 3.8.3 Procedure - Obtain the tare weight of each of the four covered petri dishes to the nearest 0.01 g. Place  $25 \pm 1$  mL of brake fluid in each dish, replace proper covers and reweigh. Determine the weight of each brake fluid test specimen by the difference. Place the four dishes, each inside its inverted cover, in the oven at  $100 \pm$  $2^{\circ}$ C for 46 ± 2 hours. (NOTE: Do not simultaneously heat more than one fluid in the same oven.) Remove the dishes from the oven, allow to cool to  $23 \pm 5^{\circ}$ C, and weigh. Return to the oven for an additional  $24 \pm 2$  hours. If at the end of  $72 \pm 4$  hours the average loss by evaporation is less than 60 percent, discontinue the evaporation procedure and proceed with examination of the residue. Otherwise, continue this procedure either until equilibrium is reached as evidenced by an incremental weight loss of less than 0.25 g in 24 hours on all individual dishes or for a maximum of 7 days. During the heating and weighing operation, if it is necessary to remove the dishes from the oven for a period of

longer than 1 hour, the dishes shall be stored in a desiccator as soon as cooled to room temperature. Calculate the percentage of fluid evaporated from each dish. Examine the residue in the dishes at the end of 1 hour at  $23 \pm 5^{\circ}$ C. Rub any sediment with the finger-tip to determine grittiness or abrasiveness. Combine the residues from all four dishes in a 4-ounce oil sample bottle and store vertically in a cold chamber at  $-5 \pm 1^{\circ}$ C for  $60 \pm 10$  minutes. Quickly remove the bottle and place in the horizontal position. The residue must flow at least 5 mm along the tube within 5 seconds.

3.8.4 Calculation - The average of the percentage evaporated from all four dishes is the loss by evaporation.

## **3.9 WATER TOLERANCE**

Evaluate the water tolerance characteristics of a brake fluid by running one test specimen according to the following procedure.

- 3.9.1 Summary of the Procedure Brake fluid, except DOT 5 SBBF, is diluted with 3.5 percent water (DOT 5 SBBF is humidified), and then it is stored at -40°C for 120 hours. The cold, water-wet fluid is first examined for clarity, stratification, and sedimentation, then placed in an oven at  $60^{\circ}$ C for 24 hours. On removal, it is again examined for stratification, and the volume percent of sediment determined by centrifuging.
- 3.9.2 Apparatus
  - a) Centrifuge tube. See 4.5.1(a).
  - b) Centrifuge. See 4.5.1(b).
  - c) Cold Chamber. See 3.7.2(b).
  - d) Oven. Gravity or forced convection oven. See 3.6.2(c).
  - e) Timing device. See 3.3.2(e).
- 3.9.3 Procedure
  - a) At low temperature. Humidify  $100 \pm 1 \text{ mL of DOT 5 SBBF in}$ accordance with 3.2, eliminating determination of the ERBP. When testing brake fluids, except DOT 5 SBBF, mix  $3.5 \pm 0.1 \text{ mL}$ of distilled water with  $100 \pm 1 \text{ mL of brake fluid; pour into a}$ centrifuge tube. Stopper the tube with a clean cork and place in

the cold chamber maintained at  $-40 \pm 2^{\circ}$ C. After  $120 \pm 2$  hours, remove tube, quickly wipe with a clean lint-free cloth saturated with ethanol (isopropanol when testing DOT 5 SBBF) or acetone, and examine the fluid for evidence of sludging, crystallization, stratification or sedimentation. Invert the tube and determine the number of seconds required for the air bubble to travel to the top of the fluid. (The air bubble is considered to have reached the top of the fluid when the top of the bubble reaches the 2 mL graduation of the centrifuge tube.) If the wet fluid has become cloudy, warm to  $23 \pm 5^{\circ}$ C and note appearance and fluidity.

b) At  $60^{\circ}$ C. Place tube and brake fluid from 3.9.3(a) in an oven maintained at  $60 \pm 2^{\circ}$ C for  $24 \pm 2$  hours. Remove the tube and immediately examine the contents for evidence of stratification. Determine the percent sediment by centrifuging as described in 4.5.

## **3.10** COMPATIBILITY

The compatibility of a brake fluid with other brake fluids shall be evaluated by running one test sample according to the following procedure.

- 3.10.1 Summary of the Procedure Brake fluid is mixed with an equal volume of SAE Compatibility Fluid, then tested in the same way as for water tolerance (see 3.9.3) except that the bubble flow time is not measured. This test is an indication of the compatibility of the test fluid with other motor vehicle brake fluids at both high and low temperatures.
- 3.10.2 Apparatus and Materials
  - a) Centrifuge tube. See 4.5.1(a).
  - b) Centrifuge. See 4.5.1(b).
  - c) Cold chamber. See 3.7.2(b).
  - d) Oven. See 3.6.2(c).
  - e) SAE Compatibility Fluid.
- 3.10.3 Procedure
  - a) At low temperature. Mix  $50 \pm 0.5$  mL of brake fluid with  $50 \pm 0.5$  mL of SAE Compatibility Fluid. Pour this mixture into a centrifuge tube and stopper with a clean dry cork. Place tube in

the cold chamber maintained at  $-40 \pm 2^{\circ}$ C. After  $24 \pm 2$  hours, remove tube, quickly wipe with a clean lint-free cloth saturated with ethanol (isopropanol when testing DOT 5 SBBFs) or acetone. Examine the test specimen for evidence of sludging, crystallization or sedimentation. Test fluids, except DOT 5 SBBF, shall also be examined for stratification.

b) At 60°C. Place tube and test fluid from 3.10.3(a) for  $24 \pm 2$  hours in an oven maintained at  $60 \pm 2$ °C. Remove the tube and immediately examine the contents of the test mixtures, except DOT 5 SBBFs, for evidence of stratification. Determine the percent sediment by centrifuging as described in 4.5.

### **3.11 RESISTANCE TO OXIDATION**

The stability of a brake fluid under oxidative conditions shall be evaluated by running duplicate samples according to the following procedure.

- 3.11.1 Summary of the Procedure Brake fluids, except DOT 5 SBBF, are activated with a mixture of approximately 0.2 percent benzoyl peroxide and 5 percent water. DOT 5 SBBF is humidified in accordance with 3.2, eliminating determination of the ERBP, and then approximately 0.2 percent benzoyl peroxide is added. A corrosion test strip assembly consisting of cast iron and an aluminum strip separated by tinfoil squares at each end is then rested on a piece of SBR WC cup positioned so that the test strip is half immersed in the fluid, and oven-aged at 70°C for 168 hours. At the end of this period, the metal strips are examined for pitting, etching, and weight loss.
- 3.11.2 Equipment
  - a) Balance. See 3.6.2(a).
  - b) Desiccators. See 3.6.2(b).
  - c) Oven. See 3.6.2(c).
  - d) Three glass test tubes approximately 22 mm outside diameter by 175 mm in length.
- 3.11.3 Reagents and Materials
  - a) Benzoyl peroxide, reagent grade, 96 percent. (Benzoyl peroxide that is brownish, or dusty, or has less than 90 percent purity, must be discarded.) Reagent strength may be evaluated by ASTM

standard E298-68, "Standard Methods for Assay of Organic Peroxides".

- b) Corrosion test strips. Two sets of cast iron and aluminum metal test strips as described in Appendix C of SAE Standard J1703 NOV83, "Motor Vehicle Brake Fluid," November 1983.
- c) Tinfoil. Four unused pieces of tinfoil approximately 12 mm square and between 0.02 and 0.06 mm in thickness. The foil shall be at least 99.9 percent tin and contain not more than 0.025 percent lead.
- d) SBR cups. Two unused, approximately one-eighth sections of a standard SAE SBR WC cup (as described in 4.6).
- e) Machine screw and nut. Two clean oil-free, No. 6 or 8-32 x 3/8 or 1/2-inch long (or equivalent metric size), round or fillister head, uncoated mild steel machine screws, with matching plain nuts.

#### 3.11.4 Preparation

- a) Corrosion test strips. Prepare two sets of aluminum and cast iron test strips according to 3.6.4(a) except for assembly. Weigh each strip to the nearest 0.1 mg and assemble a strip of each metal on a machine screw, separating the strips at each end with a piece of tinfoil. Tighten the nut enough to hold both pieces of foil firmly in place.
- b) Test mixture. Place  $30 \pm 1$  mL of the brake fluid under test in a 22 by 175 mm test tube. For all brake fluids, except DOT 5 SBBF, add  $0.060 \pm 0.002$  g of benzoyl peroxide and  $1.50 \pm 0.05$  mL of distilled water. For DOT 5 SBBFs, use test fluid humidified in accordance with 3.2 and add only the benzoyl peroxide. Stopper the tube loosely with a clean dry cork, shake and place in an oven for 2 hours at  $70 \pm 2^{\circ}$ C. Shake every 15 minutes to effect solution of the peroxide, but do not wet cork. Remove the tube from the oven and allow to cool to  $23 \pm 5^{\circ}$ C. Begin testing according to 3.11.5 not later than 24 hours after removal of the tube from the oven.
- 3.11.5 Procedure Place a one-eighth SBR cup section in the bottom of each tube. Add 10 mL of prepared test mixture to each test tube. Place a metal-strip assembly in each, the end of the strip without the screw resting on the rubber, and the solution covering about one-half the length of the strips. Stopper the tubes with clean dry corks and store upright for  $70 \pm 2$  hours at  $23 \pm 5^{\circ}$ C. Loosen the corks and place the tubes for  $168 \pm 2$  hours in an oven maintained at  $70 \pm 2^{\circ}$ C. Afterwards remove and disassemble strips. Examine

the strips and note any gum deposits. Wipe the strips with a clean cloth wet with ethanol (isopropanol when testing DOT 5 SBBFs) and note any pitting, etching or roughening of surface disregarding stain or discoloration. Place the strips in a desiccator over silica gel or other suitable desiccant, at  $23 \pm 5^{\circ}$ C for at least 1 hour. Again weigh each strip to the nearest 0.1 mg.

3.11.6 Calculation - Determine corrosion loss by dividing the change in weight of each metal strip by the total surface area of each strip measured in square centimetres, to the nearest square centimetre. Average the results for the two strips of each type of metal, rounding to the nearest 0.05 mg per square centimetre. If only one of the duplicates fails for any reason, run a second set of duplicate samples. Both repeat samples shall meet all oxidation resistance requirements of section 116 of the said Regulations.

## **3.12 EFFECT ON SBR CUPS**

The effects of a brake fluid in swelling, softening, and otherwise affecting standard SBR WC cups shall be evaluated by the following procedure.

- 3.12.1 Summary of the Procedure Four standard SAE SBR WC cups are measured and their hardnesses determined. The cups, two to a jar, are immersed in the test fluid. One jar is heated for 70 hours at 70°C, and the other for 70 hours at 120°C. Afterwards, the cups are washed, examined for disintegration, remeasured and their hardness redetermined.
- 3.12.2 Equipment and Supplies
  - a) Oven. See 3.6.2(c).
  - b) Glass jars and lids. Two screw-top, straight-sided round glass jars, each having a capacity of approximately 250 mL and inner dimensions of approximately 125 mm in height and 50 mm in diameter, and a tinned steel lid (no insert or organic coating).
  - c) SBR cups. See 4.6.
- 3.12.3 Preparation Measure the base diameters of the SBR cups as described in 3.6.4(b), and the hardness of each as described in 4.4.
- 3.12.4Procedure Wash the cups in 90 percent ethanol (isopropanol<br/>when testing DOT 5 SBBFs) (see 4.3), for not longer than<br/>30 seconds and quickly dry with a clean, lint-free cloth. Using<br/>forceps, place two cups into each of the two jars; add 75 mL of

brake fluid to each jar and cap tightly. Place one jar in an oven held at  $70 \pm 2^{\circ}$ C for  $70 \pm 2$  hours. Place the other jar in an oven held at  $120 \pm 2^{\circ}$ C for  $70 \pm 2$  hours. Allow each jar to cool for 60 to 90 minutes at  $23 \pm 5^{\circ}$ C. Remove cups, wash with ethanol (isopropanol when testing DOT 5 SBBFs) for not longer than 30 seconds, and quickly dry. Examine the cups for disintegration as evidenced by stickiness, blisters, or sloughing. Measure the base diameter and hardness of each cup within 15 minutes after removal from the fluid.

- 3.12.5 Calculation
  - a) Calculate the change in base diameter of each cup. If the two values, at each temperature, do not differ by more than 0.10 mm, average them to the nearest 0.02 mm. If the two values differ by more than 0.10 mm, repeat the test at the appropriate temperature and average the four values as the change in base diameter.
  - b) Calculate the change in hardness for each cup. The average of the two values for each pair is the change in hardness.
  - c) Note disintegration as evidenced by stickiness, blisters or sloughing.

#### **3.13 STROKING PROPERTIES**

Evaluate the lubricating properties, component compatibility, resistance to leakage and related qualities of a brake fluid by running one sample according to the following procedures.

3.13.1 Summary of the Procedure - Brake fluid is stroked under controlled conditions at an elevated temperature in a simulated motor vehicle hydraulic braking system consisting of three slave wheel cylinders and an actuating master cylinder connected by steel tubing. Referee standard parts are used. All parts are carefully cleaned, examined, and certain measurements are made immediately prior to assembly for test. During the test, temperature, rate of pressure rise, maximum pressure and rate of stroking are specified and controlled. The system is examined periodically during stroking to assure that excessive leakage of fluid is not occurring. Afterwards, the system is torn down. Metal parts and SBR cups are examined and remeasured. The brake fluid and any resultant sludge and debris are collected, examined and tested.

- 3.13.2 Apparatus and Equipment Either the drum and shoe type of stroking apparatus (see Figure 1 of SAE Standard J1703b) except using only three sets of drum and shoe assemblies, or the stroking fixture type apparatus as shown in Figure 2 of SAE J1703 NOV83, with the components arranged as shown in Figure 1 of SAE J1703 NOV83. The following components are required.
  - a) Brake assemblies. With the drum and shoe apparatus: three drum and shoe assembly units (SAE RM-29a) consisting of three forward brake shoes and three reverse brake shoes with linings and three front wheel brake drum assemblies with assembly component parts. With stroking fixture type apparatus: three fixture units including appropriate adapter mounting plates to hold brake wheel cylinder assemblies.
  - b) Braking pressure actuation mechanism. An actuating mechanism for applying a force to the master cylinder pushrod without side thrust. The amount of force applied by the actuating mechanism shall be adjustable and capable of applying sufficient thrust to the master cylinder to create a pressure of at least 6,867 kPa in the simulated brake system. A hydraulic gauge or pressure recorder, having a range of at least 0 to 6,867 kPa, shall be installed between the master cylinder and the brake assemblies and shall be provided with a shutoff valve and with a bleeding valve for removing air from the connecting tubing. The actuating mechanism shall be designed to permit adjustable stroking rates of approximately 1,000 strokes per hour. Use a mechanical or electrical counter to record the total number of strokes.
  - c) Heated air bath cabinet. An insulated cabinet or oven having sufficient capacity to house the three mounted brake assemblies or stroking fixture assemblies, master cylinder, and necessary connections. A thermostatically controlled heating system is required to maintain a temperature of  $70 \pm 5^{\circ}$ C or  $120 \pm 5^{\circ}$ C. Heaters shall be shielded to prevent direct radiation to wheel or master cylinder.
  - Master cylinder (MC) assembly (SAE RM-15a). One cast iron housing hydraulic brake system cylinder having a diameter of approximately 28 mm and fitted for a filler cap and standpipe (see 3.13.2(e)). The MC piston shall be made from SAE CA 360 copper-base alloy (half hard). A new MC assembly is required for each test.
  - e) Filler cap and standpipe. MC filler cap provided with a glass or uncoated steel standpipe. Standpipe must provide adequate

volume for thermal expansion, yet permit measurement and adjustment of the fluid level in the system to  $\pm$  3 mL. Cap and standpipe may be cleaned and reused.

- f) Wheel cylinder (WC) assemblies (SAE RM-14a). Three unused cast iron housing straight bore hydraulic brake WC assemblies having diameters of approximately 28 mm for each test. Pistons shall be made from unanodized SAE AA 2024 aluminum alloy.
- g) Micrometer. Same as 3.6.2(d).

#### 3.13.3 Materials

- a) Standard SBR brake cups. Six standard SAE SBR wheel cylinder test cups, one primary MC test cup and one secondary MC test cup, all as described in 4.6, for each test.
- b) Steel tubing. Double wall steel tubing meeting SAE Standard J527. A complete replacement of tubing is essential when visual inspection indicates any corrosion or deposits on inner surface of tubing. Tubing from master cylinder to one wheel cylinder shall be replaced for each test (minimum length 91 cm). Uniformity in tubing size is required between master cylinder and wheel cylinder. The standard master cylinder has two outlets for tubing, both of which must be used.
- 3.13.4 Preparation of Test Apparatus
  - Wheel cylinder assemblies. Use unused wheel cylinder a) assemblies. Disassemble cylinders and discard cups. Clean all metal parts with ethanol (isopropanol when testing DOT 5 SBBFs). Inspect the working surfaces of all metal parts for scoring, galling, or pitting and cylinder bore roughness, and discard all defective parts. Remove any stains on cylinder walls with crocus cloth and ethanol (isopropanol when testing DOT 5 SBBFs). If stains cannot be removed, discard the cylinder. Measure the internal diameter of each cylinder at a location approximately 19 mm from each end of the cylinder bore, taking measurements in line with the hydraulic inlet opening and at right angles to this centreline. Discard the cylinder if any of these four readings exceeds the maximum or minimum limits of 28.66 to 28.60 mm. Measure the outside diameter of each piston at two points approximately 90° apart. Discard any piston if either reading exceeds the maximum or minimum limits of 28.55 to 28.52 mm. Select parts to ensure that the clearance between each piston and mating cylinder is within 0.08 to 0.13 mm. Use unused

SBR cups. To remove dirt and debris, rinse the cups in 90 percent ethyl alcohol for not more than 30 seconds and wipe dry with a clean lint-free cloth. Discard any cups showing defects such as cuts, molding flaws, or blisters. Measure the lip and base diameters of all cups with an optical comparator or micrometer to the nearest 0.02 mm along the centreline of the SAE and rubbertype identifications and at right angles to this centreline. Determine base diameter measurements at least 0.4 mm above the bottom edge and parallel to the base of the cup. Discard any cup if the two measured lip or base diameters differ by more than 0.08 mm. Average the lip and base diameters of each cup. Determine the hardness of all cups according to 4.4. Dip the rubber and metal parts of wheel cylinders, except housing and rubber boots, in the fluid to be tested and install them in accordance with the manufacturer's instructions. Manually stroke the cylinders to ensure that they operate easily. Install cylinders in the simulated brake system.

b)

Master cylinder assembly. Use an unused master cylinder and unused standard SBR primary and secondary MC cups which have been inspected, measured and cleaned in the manner specified in 3.13.4(a), omitting hardness of the secondary MC cup. However, prior to determining the lip and base diameters of the secondary cup, dip the cup in test brake fluid, assemble on the MC piston, and maintain the assembly in a vertical position at  $23 \pm 5$ °C for at least 12 hours. Inspect the relief and supply ports of the master cylinder; discard the cylinder if ports have burrs or wire edges. Measure the internal diameter of the cylinder at two locations (approximately midway between the relief and supply ports and approximately 19 mm beyond the relief port toward the bottom or discharge end of the bore), taking measurements at each location on the vertical and horizontal centreline of the bore. Discard the cylinder if any reading exceeds the maximum or minimum limits of 28.65 to 28.57 mm. Measure the outside diameter of each end of the master cylinder piston at two points approximately 90° apart. Discard the piston if any of these four readings exceed the maximum or minimum limits of 28.55 to 28.52 mm. Dip the rubber and metal parts of the master cylinder, except the housing and push rod-boot assembly, in the brake fluid and install in accordance with manufacturer's instructions. Manually stroke the master cylinder to ensure that it operates easily. Install the master cylinder in the simulated brake system.

c)

Assembly and adjustment of test apparatus.

- (i) When using a shoe and drum type apparatus, adjust the brake shoe toe clearances to  $1.0 \pm 0.1$  mm. Fill the system with brake fluid, bleeding all wheel cylinders and the pressure gauge to remove entrapped air. Operate the actuator manually to apply a pressure greater than the required operating pressure and inspect the system for leaks. Adjust the actuator and/or pressure relief valve to obtain pressure of  $6,867 \pm 343$  kPa. A smooth pressure-stroke pattern is required when using a shoe and drum type apparatus. The pressure is relatively low during the first part of the stroke and then builds up smoothly to the maximum stroking pressure at the end of the stroke, to permit the primary cup to pass the compensating hole at a relatively low pressure. Using stroking fixtures, adjust the actuator and/or pressure of  $6,867 \pm 343$  kPa.
- (ii) Adjust the stroking rate to  $1,000 \pm 100$  strokes per hour. Record the fluid level in the master cylinder standpipe.
- 3.13.5 Procedure - Operate the system for  $16,000 \pm 1,000$  cycles at  $23 \pm$  $5^{\circ}$ C. Repair any leakage, readjust the brake shoe clearances, and add fluid to the master cylinder standpipe to bring to the level originally recorded, if necessary. Start the test again and raise the temperature of the cabinet within  $6 \pm 2$  hours to  $120 \pm 5$  °C. During the test observe operation of wheel cylinders for improper functioning and record the amount of fluid required to replenish any loss, at intervals of 24,000 strokes. Stop the test at the end of 85,000 total recorded strokes. These totals shall include the number of strokes during operation at  $23 \pm 5^{\circ}$ C and the number of strokes required to bring the system to the operating temperature. Allow equipment to cool to room temperature. Examine the wheel cylinders for leakage. Stroke the assembly an additional 100 strokes, examine wheel cylinders for leakage and record volume loss of fluid. Within 16 hours after stopping the test, remove the master and wheel cylinders from the system, retaining the fluid in the cylinders by immediately capping or plugging the ports. Disassemble the cylinders, collecting the fluid from the master cylinder and wheel cylinders in a glass jar. When collecting the stroked fluid, remove all residue which has deposited on rubber and metal internal parts by rinsing and agitating such parts in the stroked fluid and using a soft brush to assure that all loose adhering sediment is collected. Clean SBR cups in ethanol (isopropanol when testing DOT 5 SBBFs) and dry. Inspect the cups for stickiness, scuffing, blistering, cracking, chipping, and change in shape from original appearance. Within 1 hour after disassembly, measure the lip and base diameters of each cylinder cup by the procedures specified in 3.13.4(a) and (b) with

the exception that lip or base diameters of cups may now differ by more than 0.08 mm. Determine the hardness of each cup according to 4.4. Note any sludge or gel present in the test fluid. Within 1 hour after draining the cylinders, agitate the fluid in a glass jar to suspend and uniformly disperse sediment and transfer a 100 mL portion of this fluid to a centrifuge tube and determine percent sediment as described in 4.5. Allow the tube and fluid to stand for 24 hours, recentrifuge and record any additional sediment recovered. Inspect cylinder parts, note any gumming or any pitting on pistons and cylinder walls. Disregard staining or discoloration. Rub any deposits adhering to cylinder walls with a clean soft cloth wetted with ethanol (isopropanol when testing DOT 5 SBBFs) to determine abrasiveness and removability. Clean cylinder parts in ethanol (isopropanol when testing DOT 5 SBBFs) and dry. Measure and record diameters of pistons and cylinders according to 3.13.4(a) and (b). Repeat the test if mechanical failure occurs that may affect the evaluation of the brake fluid.

#### 3.13.6 Calculation

- a) Calculate the changes in diameters of cylinders and pistons.
- b) Calculate the average decrease in hardness of the seven cups tested, as well as the individual values.
- c) Calculate the increases in base diameters of the eight cups.
- d) Calculate the lip diameter interference set for each of the eight cups by the following formula and average the eight values.

 $(D_1 - D_2) / (D_1 - D_3) \times 100\%$  = percentage Lip Diameter Interference Set

Where:

 $D_1 = Original lip diameter.$  $D_2 = Final lip diameter.$ 

 $D_3 = Original$  cylinder bore diameter.

## 4. AUXILIARY TEST METHODS AND REAGENT STANDARDS

#### 4.1 DISTILLED WATER

Non-referee reagent water as specified in ASTM standard D1193-70, "Standard Specifications for Reagent Water" or water of equal purity.

#### 4.2 WATER CONTENT OF MOTOR VEHICLE BRAKE FLUIDS

Use analytical methods based on ASTM standard D1123-59, "Standard Method of Test for Water in Concentrated Engine Antifreezes by the Iodine Reagent Method," for determining the water content of brake fluids, or other methods of analysis yielding comparable results. To be acceptable for use, such other method must measure the weight of water added to samples of the SAE Compatibility Fluid and TEGME referee material within  $\pm$  15 percent of the water added for additions up to 0.8 percent by weight, and within  $\pm$  5 percent of the water added for additions greater than 0.8 percent by weight. The SAE Compatibility Fluid used to prepare the samples must have an original ERBP of not less than 205°C when tested in accordance with 3.1. The SAE TEGME fluid used to prepare the samples must have an original ERBP of not less than 240°C when tested in accordance with 3.1.

#### 4.3 ETHANOL

95 percent (190 proof) ethyl alcohol, USP or ACS, or Formula 3-A Specially Denatured Alcohol of the same concentration (as specified at Title 27, Code of Federal Regulations 21.35). For pretest washings of equipment use approximately 90 percent ethyl alcohol, obtained by adding 5 parts of distilled water to 95 parts of ethanol.

#### 4.4 MEASURING THE HARDNESS OF SBR BRAKE CUPS

Hardness measurements on SBR wheel cylinder cups and master cylinder primary cups shall be made by using the following apparatus and the following procedure.

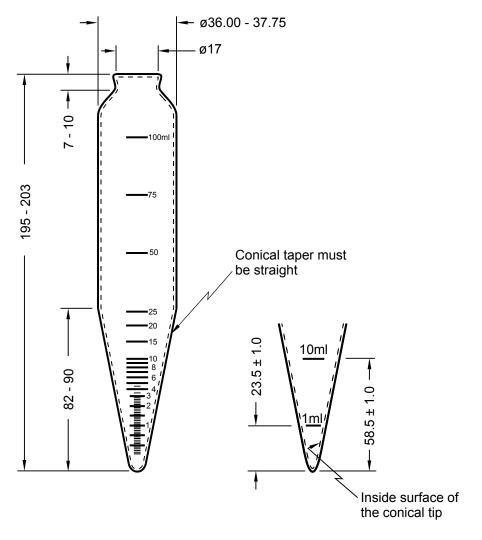
- 4.4.1 Apparatus
  - a) Anvil. A rubber anvil having a flat circular top  $20 \pm 1$  mm in diameter, a thickness of at least 9 mm and a hardness within 5 IRHDs of the SBR test cup.
  - b) Hardness tester. A hardness tester meeting the requirements for the standard instrument as described in ASTM standard D 1415-68, "Standard Method of Test for International Hardness of Vulcanized Natural and Synthetic Rubbers", and graduated directly in IRHD units.
- 4.4.2 Procedure Make hardness measurements at  $23 \pm 2^{\circ}$ C. Equilibrate the tester and anvils at this temperature prior to use. Centre brake cups lip side down on an anvil of appropriate hardness. Following the manufacturer's operating instructions for the hardness tester, make one measurement at each of four points 6 mm from the

centre of the cup and spaced  $90^{\circ}$  apart. Average the four values, and round off to the nearest IRHD.

#### 4.5 SEDIMENT BY CENTRIFUGING

The amount of sediment in the test fluid shall be determined by the following procedure.

- 4.5.1 Apparatus
- a) Centrifuge tube. Cone-shaped centrifuge tubes conforming to the dimensions given in Figure 6, and made of thoroughly annealed glass. The graduations shall be numbered as shown in Figure 6, and shall be clear and distinct. Scale-error tolerances and smallest graduations between various calibration marks are given in Table III and apply to calibrations made with air-free water at 20°C.



Notes:

1. Dimensions in mm unless specified otherwise.

2. Not to scale

#### FIGURE 6 – ASTM 8-INCH CENTRIFUGE TUBE

b)

Range. mL	Subdivision, mL	Volume tolerance, mL
0 to 0.1	0.05	± 0.02
>0.1 to 0.3	0.05	± 0.03
>0.3 to 0.5	0.05	± 0.05
>0.5 to 1	0.10	± 0.05
>1 to 2	0.10	± 0.10
>2 to 3	0.20	± 0.10
>3 to 5	0.5	± 0.20
>5 to 10	1	± 0.50
>10 to 25	5	± 1.00
>25 to 100	25	± 1.00

TABLE III - CALIBRATION TOLERANCES FOR 8-INCH CENTRIFUGE TUBE

Centrifuge. A centrifuge capable of whirling two or more filled centrifuge tubes at a speed which can be controlled to give a relative centrifugal force (r.c.f.) between 600 and 700 at the tip of the tubes. The revolving head, trunnion rings, and trunnion cups, including the rubber cushion, shall withstand the maximum centrifugal force capable of being delivered by the power source. The trunnion cups and cushions shall firmly support the tubes when the centrifuge is in motion. Calculate the speed of the rotating head using the equation:

r.p.m. = 265 x  $(r.c.f./d)^{\frac{1}{2}}$ 

where:

r.c.f. = Relative centrifugal force, and

d = Diameter of swing, in inches, measured between tips of opposite tubes when in rotating position.

Table IV shows the relationship between diameter of swing, relative centrifugal force (r.c.f.), and revolutions per minute.

4.5.2 Procedure - Balance the corked centrifuge tubes with their respective trunnion cups in pairs by weight on a scale, according to the centrifuge manufacturer's instructions, and place them on opposite sides of the centrifuge head. Use a dummy assembly when one sample is tested. Then whirl them for 10 minutes, at a rate sufficient to produce a r.c.f. between 600 and 700 at the tips of the whirling tubes. Repeat until the volume of sediment in each tube remains constant for three consecutive readings.

Diameter of swing, inches	R.p.m. at 600 r.c.f.	R.p.m. at 700 r.c.f.
19	1490	1610
20	1450	1570
21	1420	1530
22	1390	1500

#### TABLE IV – ROTATION SPEEDS FOR CENTRIFUGES OF VARIOUS DIAMETERS

Measured in inches between tips of opposite tubes when in rotating position.

# 4.6 STANDARD STYRENE-BUTADIENE RUBBER (SBR) BRAKE CUPS

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SBR brake cups for testing motor vehicle brake fluids shall be manufactured using the following formulation:

<sup>4.5.3</sup> Calculations - Read the volume of the solid sediment at the bottom of the centrifuge tube and report the percent sediment by volume. Where replicate determinations are specified, report the average value.

Ingredients	Parts by weight
SBR type 1503	100
Oil furnace black (NBS 378)	40
Zinc oxide (NBS 370)	5
Sulfur (NBS 371)	0.25
Stearic Acid (NBS 372)	1
n-tertiary butyl-2-	
benzothiazole	1
sulfenamide (NBS 384)	
Symmetrical	
dibetanaphthyl-p-	1.5
phenylenediamine	
Dicumyl peroxide (40	4.5
percent on precipitated CaCO <sub>3</sub> ) <sup>2</sup>	4.5
Total	153.25

#### FORMULATION OF RUBBER COMPOUND

1 2 Philprene 1503 has been found suitable.

Use only within 90 days of manufacture and store at temperature below 27 C.

**NOTE:** The ingredients labeled (NBS ...) must have properties identical with those supplied by the U.S. National Bureau of Standards.

Compounding, vulcanization, physical properties, size of the finished cups, and other details shall be as specified in Appendix B of SAE J1703b. The cups shall be used in testing brake fluids either within 6 months from date of manufacture when stored at room temperature below  $30^{\circ}$ C or within 36 months from date of manufacture when stored at temperatures below  $-15^{\circ}$ C. After removal of cups from refrigeration they shall be conditioned base down on a flat surface for at least 12 hours at room temperature in order to allow cups to reach their true configuration before measurement.

## 4.7 ISOPROPANOL

ACS or reagent grade.