Toxicologically Relevant Characterization of Carbon Nanomaterials

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Nanotoxicology

Some nanomaterial properties raise toxicity concerns

Small size

- small aerodynamic diameter
 - \rightarrow deep lung penetration
- high permeability in biological membranes
- enhanced cellular uptake

(endocytosis, phagocytosis)

High surface area

- high surface activity
- facilitated transport the "Trojan horse" effect

Fibrous morphology

- entanglement and airway blockage
- difficulty with macrophage clearance

carbon nanotubes: 100 gms!



Is there a Role for Materials Science in Nanotoxicology?

Carbon nanotubes are a *family* of complex materials



CNTs vary greatly in:

- diameter (0.4 nm 200 nm) and length
- surface chemistry (hydrophilic / phobic)
- coatings (polymers, surfactants, proteins, DNA).
- aggregation state
- composition (Fe, Co, Y, Ni, amorphous carbon, graphite)



Thess et al. Science, 1996







Basic nanomaterial properties relevant to toxicity

- > Dose
- > Size and shape
- > Biopersistence
- > Surface chemistry



Surface Reactivity of Nanoparticles *Nel et al. Science 311: 622-627, 2006*

Suggested Roles for Materials Scientists

- 1. professional characterization of materials used in toxicity testing
 - Material names "MWNTs" or "SWNTs" are not enough !

Key issue: what characterization methods are most relevant to toxicity?

2. synthesis / processing of model materials for *mechanistic* toxicology



3. development of "green" nanomaterials or eco-nanomaterials

-- the co-optimization of materials for both function and safety *

* also known as "materials safety by design"

Toxicity of Fullerenes Depends on Surface State



Possible Mechanisms of Fullerene Toxicity

based on Sayes et al., 2004, Oberdorster, 2004



Simulated lipid bilayer structure



Unsubstituted fullerene aggregates in aqueous media to form "nano-C60"

Postulated mechanism:

Hydrophobic attachment to / incorporation in cell membranes with redox catalysis of lipid peroxidation

Toxicity of Functionalized Carbon Nanotubes

Dumortier et al., Nano Letters, 2006





 Both CNT types are taken up by immune cells (B, T lymphocytes, macrophages) and are non-cytotoxic Green fluorescence



The less soluble type (below) does elicit release of pro-inflammatory cytokines from macrophages; the more soluble formulation (above) does not

Catalyst Residues in Carbon Nanotube Samples



Catalytic Synthesis of Carbon Nanotubes

hydrocarbon decomposition

diffusion

Vendor A	AP SWNT	Ni-Y (35%)
В	AP SWNT	Ni-Co (25%)
С	AP SWNT	Ni-Y (25%)
	High purity (HP)	Co-Mo (<2%)
D	AP SWNT	Ni-Y (30%)
	Purified SWNT	Ni-Y (15%)
Е	Unpurified MWNT	Fe (4.2%)
	Purified MWNT	Fe (0.1%)
F	Unpurified MWNT	Fe (4.25%)
	Purified MWNT	Fe (3.29%)
G	Unpurified SWNT	Fe (22.2%)
	Purified SWNT	Fe (10.9%)

graphene layer – precipitation



Can metals cause or contribute to CNT toxicity?

- Catalytic growth methods:
 - -- now dominant for synthesis of multiwall nanotubes (esp. large scale)
 - -- only route for single-wall nanotube synthesis
- Over 20 metals have been used to synthesize carbon nanotubes.
 Most common elements in CNT catalyst formulations are Fe, Ni, Y, Co, Mo





The nickel-ion hypothesis: nickel toxicity, carcinogenesis mainly depend on intracellular nickel(II) ion concentrations, independent of the original nickel compound [Snow, 1998].

Ni mobilized at extracellular pH (7)

Various commercial samples



Ni mobilized at lysosomal pH (5.5)

Various commercial samples



Nickel Mobilization from SWNTs: Effect of pH and Media







Ni Release into Cell Culture Medium and Uptake by Human Lung Epithelial Cells





The nickel-ion hypothesis: nickel toxicity, carcinogenesis mainly depend on intracellular nickel(II) ion concentrations, independent of the original nickel compound [Snow, 1998].

Free iron can catalyze radical generation and oxidative stress



Fe-containing CNTs

glutathione depletion

Fe/C Association Morphologies





Fe-C morphologies in commercial CNTs



Fe Release **From CNTs**

70



D 90







Asbestos
Unpurified CNT
Purified CNT
Ground CNT
Oxidized CNT



Toward Carbon Nanotube Detoxification



CELLULAR UPTAKE

Macrophages actively engulf (phagocytose) foreign bodies and transport them up the mucocilliary escalator out of the lung

Can macrophages phagocytose nanomaterials completely, and without injury or inflammatory signaling?



Macrophages adhering to large aggregate of carbon nanofibers

frustrated phagocytosis of asbestos fiber





1 um



M. Desjardins, Nature Reviews/Immunology, 3:280, 2003

Summary

Materials names / labels (SWNTs, as-produced, "purified") are of limited use. We need materials characterization and it should be <u>toxicologically-relevant</u>

Likely key variables:

- size, shape, aggregation state, and surface chemistry (hydrophobicity, redox activity)
- metals content, phase, location as contributors to bioavailability

Because most nanomaterials are:

- fabricated, not natural
- developmental, not commercial

 \rightarrow often no large barriers to reformulation

The Opportunity

Understand relationships between toxicity and specific material features (size, shape, metals content, surface chemistry) to guide the development of intrinsically safe nanomaterials



BACKUPS

TECHNICAL PROBLEMS WITH NANOTOXICITY ASSAYS

MONTEIRO-RIVIERE AND INMAN, CARBON 44: 1070-1078, 2006.

- Dose metrics mass, particle number, surface area, dose delivered to target cell or tissue
- Acute vs. chronic toxicity endpoints
- Particle interference with fluorescent or colorimetric assays
- Adsorption of serum or cellular proteins
- Nanoparticle aggregation or agglomeration







SWNTs form large aggregates in cell culture medium containing 10% serum and induce toxicity in human lung epithelial cells. Live/Dead viability cytotoxicity assay (Molecular Probes, 200X) Attempts to disperse nanoparticles: Sonication Detergents DMSO Lung surfactant Serum PEG Organic solvents

ENGINEERING SOLUTIONS FOR NANOTOXICOLOGY ASSAYS



10 um



SEM micrographs of 1-5 um SWNT aggregates bar-coated on to a quartz substrate from ethanol suspension



Bundles of SWNTs linked to aggregates of carbon-coated catalyst nanoparticles

Robert Hurt and Lorin Jakubek, Biomedical engineering graduate student